

Naming Compounds Problems11-Sept-2009**Sodium Chloride**

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**Potassium Iodide**

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**Calcium Sulfide**

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**Cesium Bromide**

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**Magnesium Oxide**

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**Cobalt (III) Chloride**

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**Copper (I) Iodide**

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**Tin (IV) Bromide**

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**Mercury (II) Chloride**

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**Lead (II) Sulfide**

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**Potassium Nitride**

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**Mercury (II) Oxide**

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**Rubidium Fluoride**

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**Sodium Hydride**

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**Chromium (II) Fluoride**

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**Magnesium Bromide**

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**Manganese (II) iodide**

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**Lithium Oxide**

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**Diiodine Heptoxide**

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**Carbon Dioxide**

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**Carbon Tetrafluoride**

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**Ammonia**

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**Ammonium Hydroxide**

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**Phosphorous Trichloride**

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**Carbon Monoxide**

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**Nitrogen Dioxide**

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**Selenium Hexafluoride**

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**Silicon Dioxide**

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**DiHydrogen Monoxide**

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**Aluminum TriChloride**

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NaClO	_____
NaClO <sub>2</sub>	_____
KClO <sub>3</sub>	_____
KClO <sub>4</sub>	_____
(NH <sub>4</sub> ) <sub>2</sub> CO <sub>3</sub>	_____
NH <sub>4</sub> NO <sub>2</sub>	_____
NH <sub>4</sub> NO <sub>3</sub>	_____
K <sub>2</sub> SO <sub>3</sub>	_____
Na <sub>2</sub> SO <sub>4</sub>	_____
NaHSO <sub>3</sub>	_____
NaHSO <sub>4</sub>	_____
K <sub>2</sub> CO <sub>3</sub>	_____
NaHCO <sub>3</sub>	_____
H <sub>2</sub> SO <sub>4</sub>	_____
H <sub>2</sub> SO <sub>3</sub>	_____
HI	_____
HF	_____
HNO <sub>3</sub>	_____
HNO <sub>2</sub>	_____

H<sub>2</sub>SO<sub>3</sub> \_\_\_\_\_

FeCl<sub>3</sub> \_\_\_\_\_

HF \_\_\_\_\_

H<sub>2</sub>SO<sub>4</sub> \_\_\_\_\_

HCl \_\_\_\_\_

CuCl \_\_\_\_\_

HBr \_\_\_\_\_

HNO<sub>2</sub> \_\_\_\_\_

OF<sub>2</sub> \_\_\_\_\_

ClF<sub>3</sub> \_\_\_\_\_

HI \_\_\_\_\_

HNO<sub>3</sub> \_\_\_\_\_

MnO<sub>2</sub> \_\_\_\_\_

HCN \_\_\_\_\_

H<sub>3</sub>PO<sub>4</sub> \_\_\_\_\_

HgO \_\_\_\_\_

FeCl<sub>2</sub> \_\_\_\_\_

PCl<sub>5</sub> \_\_\_\_\_

H<sub>2</sub>S \_\_\_\_\_

MgI<sub>2</sub> \_\_\_\_\_

P<sub>4</sub>O<sub>6</sub> \_\_\_\_\_

SF<sub>6</sub> \_\_\_\_\_

Rb<sub>2</sub>O \_\_\_\_\_

HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> \_\_\_\_\_

NH<sub>3</sub> \_\_\_\_\_

SO<sub>3</sub> \_\_\_\_\_

CoBr<sub>2</sub> \_\_\_\_\_

SrI<sub>2</sub> \_\_\_\_\_

K<sub>2</sub>S \_\_\_\_\_

SO<sub>2</sub> \_\_\_\_\_

SrO \_\_\_\_\_

Br<sub>2</sub>O<sub>3</sub> \_\_\_\_\_

CaCl<sub>2</sub> \_\_\_\_\_

Al<sub>2</sub>O<sub>3</sub> \_\_\_\_\_

PbBr<sub>2</sub> \_\_\_\_\_

PbBr<sub>4</sub> \_\_\_\_\_

CsF \_\_\_\_\_

AlCl<sub>3</sub> \_\_\_\_\_

FeS \_\_\_\_\_

Ti Cl<sub>4</sub> \_\_\_\_\_

K<sub>2</sub>S \_\_\_\_\_

Hg<sub>2</sub>O \_\_\_\_\_

Fe<sub>2</sub>S<sub>3</sub> \_\_\_\_\_

Fe<sub>2</sub>O<sub>3</sub> \_\_\_\_\_

CuO \_\_\_\_\_

AlBr<sub>3</sub> \_\_\_\_\_

Na<sub>2</sub>S \_\_\_\_\_

CoCl<sub>3</sub> \_\_\_\_\_

BF<sub>3</sub> \_\_\_\_\_

CCl<sub>4</sub> \_\_\_\_\_

NO \_\_\_\_\_

NO<sub>2</sub> \_\_\_\_\_

N<sub>2</sub>O<sub>3</sub> \_\_\_\_\_

IF<sub>5</sub> \_\_\_\_\_

CO \_\_\_\_\_

CO<sub>2</sub> \_\_\_\_\_

H<sub>2</sub>O \_\_\_\_\_

PbCl<sub>4</sub> \_\_\_\_\_

VF<sub>5</sub> \_\_\_\_\_

CuCl \_\_\_\_\_

PbO<sub>2</sub> \_\_\_\_\_

O<sub>2</sub>F<sub>2</sub> \_\_\_\_\_

MnO<sub>2</sub> \_\_\_\_\_

MgO \_\_\_\_\_

HOH \_\_\_\_\_

XeF<sub>6</sub> \_\_\_\_\_